# HYDROCARBONS FROM CIRSIUM JAPONICUM

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Abstract—Aplotaxene, dihydro-, tetrahydro- and hexahydro- aplotaxene, 1-pentadecene, cyperene, caryophyllene, thujopsene and  $\alpha$ -himachalene were found in the root oil of Cirsium japonicum. The isolation of dihydro- and tetrahydro-aplotaxene is the first report from natural sources. The latter is a biogenetically important intermediate from oleic acid to C<sub>1.7</sub>-acetylene compounds.

### INTRODUCTION

It has been suggested that a biogenetic pathway to acetylenic compounds from oleic acid probably exists [1] but no intermediates have been isolated from natural sources. In the work reported here, one such intermediate was isolated from the root oil of Cirsium japonicum (Noazami).

## RESULTS AND DISCUSSION

Dihydroaplotaxene (1),  $C_{17}H_{30}$ , m/e 234 (M<sup>+</sup>), showed no absorption maximum in its UV spectrum but on

2.75 ppm (2H, m, CH=CH—C $H_2$ —CH=CH), three --CH<sub>2</sub>-- groups connected to a double bond at 2.04 ppm (6H, m,  $[C=C-CH_2]_3$ ), a Me group at 0.86 ppm (3H, t, J=6 Hz,  $CH_2-CH_3$ ) and six  $CH_2$  groups at 1.33 ppm (12H, s-m,  $[CH_2]_6$ ). n-Heptadecatriene after ozonlysis, gave formaldehyde, caproic acid and pimelic acid.

From the chemical and physical data, the hydrocarbon could have two possible structure 1 or 2 but as the -CH<sub>2</sub>— protons connected to a double bond at 2.04 ppm had been decoupled and the vinyl proton at 5.82 ppm (CH=CH<sub>2</sub>) changed from multiplet signals to typical quartet signals the compound had a partial structure

Me 
$$-(CH_2)_4$$
  $-CH$   $-CH_2$   $-CH$   $-CH$ 

hydrogenation consumed 3 molar equivalents of hydrogen, and gave n-haptadecane. Thus, this compound is n-heptadecatriene. The IR bands at 3075 (CH<sub>2</sub>), 1640 (C=C), 992 (CH) and 909 (CH<sub>2</sub>) cm<sup>-1</sup> showed the presence of a vinyl group. Although a band characteric of a disubstituted ethylenic double bond was observed at 3010 cm<sup>-1</sup>, a trans-disubstituted absorption was absent in the region 980-960 cm<sup>-1</sup> but a *cis*-disubstituted absorption was observed at 721 cm<sup>-1</sup> which overlapped with the -CH<sub>2</sub>-rocking vibration. The PMR spectrum provided additional information about the structure of (1); a vinyl group at 5.82 ppm (1H, m, CH=CH<sub>2</sub>), 4.97 ppm

(1H, 
$$d-m$$
,  $J = 17$  Hz,  $C=C$ ) and 4.92 ppm (1H,  $d-m$ 

(1H, 
$$d$$
-m,  $J = 17$  Hz,  $C$ =C) and 4.92 ppm (1H,  $d$ -m,  $H$ 
 $J = 10$  Hz,  $C$ =C), two  $cis$ -disubstituted ethylenic double  $H$ 

bonds at 5.34 ppm (4H, m, J < 11 Hz, [C<u>H</u>=C<u>H</u>]<sub>2</sub>),a -CH<sub>2</sub> group situated between two double bonds at (CH<sub>2</sub>)<sub>5</sub>—CH=CH<sub>2</sub> and was thus 1, 8, 11-heptadecatriene

Tetrahydroaplotaxene (3),  $C_{17}H_{32}$ , m/e 236 (M<sup>+</sup>), also had no UV absorption maximum. On hydrogenation the hydrocarbon consumed 2 molar equivalents of hydrogen and gave n-heptadecane. This compound is thus n-heptadecadiene. The IR bands, at 3075 (CH<sub>2</sub>) 1640 (C=C), 995 (CH), 910 (CH<sub>2</sub>) cm<sup>-1</sup>, indicated the presence of a vinyl group and those at 3005 and 722 cm<sup>-1</sup> a cis disubstituted ethylenic double bond. The PMR spectrum showed signals at 5.82 ppm (1H, m, CH=CH<sub>2</sub>), 4.97 ppm

(1H, 
$$d$$
-m,  $J$  = 17 Hz, C=C) and 4.92 ppm (1H,  $d$ -m, H

H

 $J$  = 10 Hz, C=C), a  $cis$ -disubstituted ethylenic double H

$$J = 10 \text{ Hz}$$
, C=C), a *cis*-disubstituted ethylenic double

bond at 5.34 ppm (2H, m, J < 10 Hz, CH=CH), 3 -CH<sub>2</sub>- groups connected to a double bond at 2.00 ppm (6H, m, [C=C-CH<sub>2</sub>]<sub>3</sub>), a Me group at 0.86 ppm 264 K. Yano

(3H, t, J = 6 Hz,  $CH_2 - C\underline{H}_3$ ) and 9  $CH_2$  groups at 1.24 ppm (18 H, s-m,  $[C\underline{H}_2]_9$ ).

n-Heptadecadiene was ozonized, and gave HCHO pelargonic acid and pimelic acid. Thus, the structure of tetrahydroaplotaxene was found to be 1,8-heptadecadiene.

The isolation of dihydro- and tetrahydro-aplotaxene is the first report from natural sources. The latter is biogenetically important intermediate from oleic acid to  $C_{17}$ -acetylene compounds [1].

Aplotaxene [2], hexahydroaplotaxene (1-heptadecene), 1-pentadecene, cyperene, caryophyllene, thujopsene and  $\alpha$ -himachalene were also identified as constituents of the root oil.

#### **EXPERIMENTAL**

UV spectra were measured in EtOH. IR spectra were recorded as liquid films and KBr disks. NMR spectra were determined using TMS as an internal standard in CDCl<sub>3</sub>. To analyse volatile constituents and to identify the minor components, GLC was carried out using 25 % PEG 6000 (He 30 ml/min, 165°) and 5 % SE 30 (He 15 ml/min, 165°).

Extraction and separation. Fresh roots (6.3 kg) of C. japonicum, collected in the beginning of May at Fukuoka prefecture, were chopped finely and extracted with  $Et_2O$  (7.6 l.) for 2 weeks at room temp. The oil (43 g, 0.68 % of the root) was obtained by evaporating the  $Et_2O$  under red pres. The extract (43 g) was chromatographed on a deactivated  $Al_2O_3$  column. n-Hexane eluted hydrocarbons (79% of the extract),  $C_6H_6$ -EtOAc (4:1) and EtOAc eluted polar components (18% of the extract). The hydrocarbon (34 g) was rechromatographed on a deactivated  $Al_2O_3$  column with n-hexane and divided into 7 fractions.

Dihydroaplotaxene (1). Fraction 6 (11% of hydrocarbon) found: C, 87.30; H, 12.67%. Calcd for C<sub>17</sub>H<sub>30</sub>: C, 87.10; H, 12.90%. 1 (49 mg) was hydrogenated over Adams Pt<sub>2</sub>O (13 mg) in HOAc and absorbed 3.1 mol of H<sub>2</sub>. n-Heptadecane (23.2 mg) was obtained, and identified by MS and GLC comparison with authentic material. 1 (111 mg) was ozonised in HOAc (15 ml) at 0° for 1 h. The reaction mixture was diluted with H<sub>2</sub>O (30 ml) and refluxed for 1 h. The mixture was distilled into an acidified alcoholic soln acidified of 2,4-DNPH and the resultant soln chromatographed on Si gel TLC with hexane-C<sub>6</sub>H<sub>6</sub>-EtOAc (7:2:1). A spot  $(R_f, 0.18)$  was identified as HCHO 2,4-DNPH by comparison with an authentic sample. The distilled residue of the reaction mixture was extracted with 100 ml of Et<sub>2</sub>O-H<sub>2</sub>O (1:1). From the aq layer, an oily substance (25 mg) was isolated, and recrystallized from  $C_6H_6$  mp 100–102°C. IR spectrum, 1695, 1430, 1410, 1350, 1275, 1202, 920, 732, 690 cm<sup>-1</sup>, was identical with that of pimelic acid. From the Et<sub>2</sub>O layer, an oil (trace) was obtained which by TLC on Si gel with iso PrOH-HOAc (98.5:1.5), gave a spot  $(R_f \ 0.62)$  identical to that of authentic caproic acid. MS of 1 showed peaks at m/e 41 (100%) 55 (61), 67 (91), 81 (71), 95 (38), 110 (39), 234 (14, M<sup>+</sup>).

Tetrahydroaplotaxene (3). Fraction 5 (7%) found: C, 86.65; H, 13.44%. Calcd for  $C_{17}H_{32}$ : C, 86.36; H, 13.64%. 3 (58 mg) was hydrogenated over Adams  $Pt_2O$  (15 mg) in HOAc and absorbed 2.2 mol of  $H_2$ . The product (32 mg) was identified as *n*-heptadecane by MS and GLC comparison with an authentic sample. 3 (153.8 mg) was ozonized in HOAc (20 ml) at 0° for 1 h. HCHO and pimelic acid were identified as reaction products using similar methods to those used for (1). The third product

was identified as pelarganic acid by IR, MS and TLC comparison with authentic material. MS spectrum of 3 showed the fragment peaks at m/e 29 (43%), 41 (99), 43 (100), 54 (47), 55 (77), 67 (56), 81 (49), 82 (52), 96 (52), 236 (20,  $M^+$ ).

Aplotaxene. Fraction 7 (67%) found: C, 87.62; H, 12.03%. Calcd for  $C_{17}H_{28}$ ; C, 87.86; H, 12.14%.  $[\alpha]_{0}^{25}$  0°. The hydrocarbon (98.6 mg) was hydrogenated over Adams Pt<sub>2</sub>O (24.7 mg) in HOAc. The uptake of H<sub>2</sub> corresponded to 4.1 mol. The product (71 mg) was isolated as crystals mp 20.5–21.5°, Found: C, 84.96; H, 14.90%. Calcd for  $C_{17}H_{36}$ : C, 84.91; H, 15.09%. The MS was identical to that of *n*-heptadecane. Ozonolysis of *n*-heptadecatetraene gave HCHO and propionaldehyde (TLC-2,4-DNPH) and pimelic acid.

The MS of aplotaxene showed peaks at m/e 41 (32%), 55 (44), 67 (70), 79 (100), 80 (50), 91 (35), 93 (45), 95 (37), 108 (65), 232 (5,  $M^+$ ). PMR (ppm): 5.82 (1H, m, CH=CH<sub>2</sub>), 4.97 (1H, d-m,

$$J = 17 \text{ Hz}, C = C$$
), 4.92 (1H,  $d-m$ ,  $J = 10 \text{ Hz}, C = C$ ), 5.35 (6H, H

m, J < 11 Hz, cis [CH=CH]<sub>3</sub>, 2.78 (4H, t-m, J = 6 Hz, CH=CH—CH<sub>2</sub>—CH=CH—CH—CH<sub>2</sub>—CH=CH), 2.04 (6H, m, [C=C—CH<sub>2</sub>]<sub>3</sub>), 0.96 (3H, t, J = 6 Hz, CH<sub>2</sub>—CH<sub>3</sub>) 1.33 (6H, s-m, [CH<sub>2</sub>]<sub>3</sub>). The CH<sub>2</sub> protons at 2.04 ppm had been decouplied. A vinyl proton at 5.82 ppm (CH=CH<sub>2</sub>) changed from complex signals to typical quartet signals. <sup>13</sup>C—NMR spectrum 14.0, 20.3, 114.1, 139.0 ppm, showed the presence of an Et group connected to a double bond and a vinyl group. From these chemical and physical data, the hydrocarbon had a partial structure of (CH<sub>2</sub>)<sub>5</sub>—CH=CH<sub>2</sub> group, and was confirmed as 1, 8, 11, 14-heptadecatetraene. IR spectrum, 3075, 3015, 1640, 992, 910, 715 cm<sup>-1</sup>, was identical with that of aplotaxene [2].

Hexahydroplotaxene ((1-heptadecene). Fraction 3 contained hyxahydroaplotaxene (2%) which was identified by GLC and MS comparison with an authentic sample.

1-Pentadecene. Fraction 2 (9%) was 1-pentadecene which was identified by IR, GLC and MS comparison with an authentic sample.

Cyperene. Fraction 1 (3%) was identified as cyperene. PMR (ppm): 0.74 (6H, s,  $C[CH_3]_2$ , 0.77 (3H, d, J = 6 Hz,  $CH = CH_3$ ) 1.60 (3H, d, J = 1.5 Hz,  $CH = C - CH_3$ ). MS: m/e 41 (52%), 55 (37), 91 (31), 105 (39), 119 (39), 161 (41), 189 (59), 204 (100,  $M^+$ ). IR spectrum identical to lit. [3].

Caryophyllene, thujopsene and  $\alpha$ -himachalene. These compounds were identified as caryophyllene (trace, PEG 6000;  $t_R$  6.3 min, SE 30; 5.4), thujopsene (trace, 7.2, 5.4) and  $\alpha$ -himachalene (trace, 7.9, 5.6) by GLC comparison with authentic samples.

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